



REGIONAL CENTRE OF ADVANCED TECHNOLOGIES AND MATERIALS

RUDOLF ZAHRADNÍK LECTURE

General Director of RCPTM cordially invites you to the lecture
in the framework of **RUDOLF ZAHRADNÍK LECTURE SERIES**

This talk will be delivered by



Prof. Pavel Jungwirth

(Institute of Organic Chemistry
and Biochemistry, AS CR)



„Unraveling the complex nature of hydrated electrons in conventional and non-conventional ways ”

**Tuesday, April 12, 11:30 am, assembly hall of Faculty of Science,
17. listopadu 12, Olomouc.**

Pavel Jungwirth graduated in physics at the Faculty of Mathematics and Physics at Charles University in Prague, focusing on chemical physics. Prof. Jungwirth received his PhD for his work in computational chemistry, he worked under the guidance of prof. Rudolf Zahradnik. Since 2011, he is a group leader at the Institute of Organic Chemistry and Biochemistry, AS CR. His research activities are focused on molecular simulations of ions at aqueous interfaces including interactions of ions with proteins and membranes; chemistry of aqueous aerosols, structure and dynamics of solvated electrons. The goal is to understand the specific ion effects in proteins, biomembranes and in water drops.

Abstract: Interaction of water with ionizing radiation is, in addition to DNA damage, also important for nuclear waste treatment. Upon photoionization in water an electron and a cationic hole are formed, and we have followed the fate of both species. In the talk, I focus on the structure and dynamics of the hydrated electron which, despite its key role in radiative processes in water, has remained elusive. The traditional cavity model has been questioned recently, but the newly suggested picture of an electron delocalized over a region of enhanced water density is controversial. Here, we present results from ab initio molecular dynamics simulations, where not only the excess electron but also the valence electrons of the surrounding water molecules are described quantum mechanically. Unlike in previous one-electron pseudopotential calculations, many-electron interactions are explicitly accounted for. The present approach allows for partitioning of the electron solvated in liquid water into contributions from an inner cavity, neighboring water molecules, and a diffuse tail. We demonstrate that all three of these contributions are sizable and, consequently, important, which underlines the complex nature of the hydrated electron and warns against oversimplified interpretations based on pseudopotential models. We also investigate the electron solvated at the water surface. The present results clearly demonstrate that the surface electron is mostly buried in the interfacial water layer, with only about ten percent of its density protruding into the vapor phase. Consequently, it has structure which is very similar to that of an electron solvated in the aqueous bulk. Finally, we show how to follow solvated electrons upon dissolving sodium in water (yes, it is explosive!).